DomTechnical Note 103PHPH Effects on Pi's FreeChlorine Sensor

How to interpret the pH graph

The graph below shows how sensitive the Pi free chlorine sensor is to pH.

The Free Chlorine sensor shows some susceptibility to pH variations above pH 7.

As the pH increases then susceptibility increases.

Very few people need to buffer their chlorine probe. Whether they do or not will depend on what errors are acceptable to them. In practice Pi has sold over 4000 free chlorine sensors and none of them are buffered.





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pH varies by +/- 0.1pH

0.5ppm would vary by +/- 0.02ppm 0.5ppm would vary by +/- 0.02ppm 0.5ppm would vary by +/- 0.05ppm

pH varies by +/- 0.5pH

1ppm +/- would vary by +/- 0.02ppm 1ppm +/- would vary by +/- 0.04ppm 1ppm +/- would vary by +/- 0.1ppm 3ppm +/- would vary by +/- 0.06ppm 3ppm +/- would vary by +/- 0.12ppm 3ppm +/- would vary by +/- 0.3ppm

pH varies by +/- 1pH

eg. at pH 8

pH varies by +/- 0.1pH	pH varies by +/- 0.5pH	pH varies by +/- 1pH
0.5ppm would vary by +/- 0.02ppm	0.5ppm would vary by +/- 0.05ppm	0.5ppm would vary by +/- 0.08ppm
1ppm +/- would vary by +/- 0.04ppm	1ppm +/- would vary by +/- 0.1ppm	1ppm +/- would vary by +/- 0.15ppm
3ppm +/- would vary by +/- 0.12ppm	3ppm +/- would vary by +/- 0.3ppm	3ppm +/- would vary by +/- 0.45ppm

eq. at pH 9

pH varies by +/- 0.1pH

pH varies by +/- 0.5pH

pH varies by +/- 1pH

0.5ppm would vary by +/- 0.03ppm 0.5ppm would vary by +/- 0.07ppm 0.5ppm would vary by +/- 0.16 ppm 1ppm +/- would vary by +/- 0.06ppm 1ppm +/- would vary by +/- 0.14ppm 1ppm +/- would vary by +/- 0.30 ppm 3ppm +/- would vary by +/- 0.18ppm 3ppm +/- would vary by +/- 0.42ppm 3ppm +/- would vary by +/- 0.90 ppm

Notes

- These figures are approximate and may vary from probe to probe.
- The effect on the sensor is predictable so that when the pH goes up the probe signal goes down and vice versa. •
- When the pH is restored the probe will return to the original value. •
- The normally accepted accuracy of a DPD test is +/- 0.06ppm. •
- If the free chlorine sensor pH susceptibility is unacceptable to your process then there are CO_2 and acetic acid buffers available, or pH compensation using a pH sensor.





